where $M_{RL}\text{=}mass$ emissions from the running loss test (see §86.134), g.

- (d)(1) For the full three-diurnal test sequence, there are two final results to report:
- (i) The sum of the adjusted total mass emissions for the diurnal and hot soak tests ($M_{DI}+M_{HS}$); and
- (ii) The adjusted total mass emissions for the running loss test, on a grams per mile basis= M_{RL}/D_{RL} , where D_{RL} =miles driven for the running loss test (see §86.134-96(c)(6)).
- (2) For the supplemental two-diurnal test sequence, there is one final result to report: the sum of the adjusted total

mass emissions for the diurnal and hot soak tests $(M_{DI}+M_{HS})$, described in §§86.133–96(p) and 86.138–96(k), respectively.

[58 FR 16043, Mar. 24, 1993, as amended at 59 FR 48510, Sept. 21, 1994; 60 FR 34348, June 30, 1995; 60 FR 43897, Aug. 23, 1995]

§86.144-90 Calculations; exhaust emissions.

The final reported test results shall be computed by use of the following formula:

(a) For light-duty vehicles and light duty trucks:

$$Y_{wm} = 0.43 \left(\frac{Y_{ct} + Y_s}{D_{ct} + D_s} \right) + 0.57 \left(\frac{Y_{ht} + Y_s}{D_{ht} + D_s} \right)$$

Where:

- (1) $Y_{\rm wm}$ =Weighted mass emissions of each pollutant, *i.e.*, HC, CO, NO_X or CO₂, in grams per vehicle mile and if appropriate, the weighted total hydrocarbon equivalent mass in grams per vehicle mile.
- (2) Y_{ct} =Mass emissions as calculated from the "transient" phase of the cold start test, in grams per test phase.
- (3) Y_{ht}=Mass emissions as calculated from the "transient" phase of the hot start test, in grams per test phase.
- (4) Y_s=Mass emissions as calculated from the "stabilized" phase of the cold start test, in grams per test phase.
- (5) D_{ct} =The measured driving distance from the ''transient'' phase of the cold start test, in miles.
- (6) D_{ht} =The measured distance from the "transient" phase of the hot start test, in miles.
- (7) D_s =The measured driving distance from the ''stabilized'' phase of the cold start test, in miles.

- (b) The mass of each pollutant for each phase of both the cold start test and the hot start test is determined from the following:
 - (1) Hydrocarbon mass:
- $HC_{mass} = V_{mix} \times Density_{HC} \times (HC_{conc}/1,000,000)$
- (2) Oxides of nitrogen mass:
- $\begin{array}{l} NO_{xmass} = V_{mix} \times Density_{NO2} \times_{H} \times (NO_{xconc} / 1,000,000) \end{array}$
 - (3) Carbon monoxide mass:
- $CO_{mass} = V_{mix} \times Density_{CO} \times (CO_{conc}/1,000,000)$
- (4) Carbon dioxide mass:
- $CO_{2mass} = V_{mix} \times Density_{CO2} \times (CO_{2conc}/100)$
 - (5) Methanol mass:
- $CH_3OH_{mass} = V_{mix} \times Density_{CH3OH}$ ($CH_3OH_{conc}/1,000,000$)

×

- (6) Formaldehyde mass:
- $HCHO_{mass}=V_{mix}\times Density_{HCHO}\times (HCHO_{conc}/1,000,000)$
- (7) Total hydrocarbon equivalent mass:
 - (i) THCE=

$$\mathrm{HC_{Mass}} + \frac{13.8756}{32.042} \Big(\mathrm{CH_{3}OH_{mass}}\Big) + \frac{13.8756}{30.0262} \Big(\mathrm{HCHO_{mass}}\Big)$$

- (c) Meaning of symbols:
- (1)(i) HC_{mass} =Hydrocarbon emissions, in grams per test phase.

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- (ii) Density $_{HC}$ =Density of hydrocarbons is 16.33 g/ft³ (0.5768 kg/m³), assuming an average carbon to hydrogen ratio of 1:1.85, at 68 °F (20 °C) and 760 mm Hg (101.3 kPa) pressure.
- (iii)(A) HC_{conc} =Hydrocarbon concentration of the dilute exhaust sample corrected for background, in ppm carbon equivalent, *i.e.*, equivalent propane $\times 3$
 - (B) $HC_{conc}=HC_e-HC_d(1-1/DF)$.

Where:

- (iv)(A) HC_e =Hydrocarbon concentration of the dilute exhaust sample or, for diesel (or methanol-fueled vehicles, if selected), average hydrocarbon concentration of the dilute exhaust sample as calculated from the integrated HC traces, in ppm carbon equivalent.
 - (B) $HC_e = FID HC_e (r)C_{CH3OHe}$
- (v) FID HC_e =Concentration of hydrocarbon plus methanol in dilute exhaust as measured by the FID, ppm carbon equivalent.
- (vi) r=FID response to methanol.
- (vii) C_{CH3OHe} =Concentration of methanol in dilute exhaust as determined from the dilute exhaust methanol sample in ppm carbon. For vehicles not fueled with methanol, C_{CH3OHe} equals zero.
- (viii)(A) HC_d =Hydrocarbon concentration of the dilution air as measured, in ppm carbon equivalent.
 - (B) $HC_d = FID HC_d (r)C_{CH3OHd}$
- $(ix)\ FID\ HC_d=Concentration$ of hydrocarbon plus methanol in dilution air as measured by the FID, ppm carbon equivalent.
- (x) C_{CH3OHd} =Concentration of methanol in dilution air as determined from dilution air methanol sample in ppm carbon. For vehicles not fueled with methanol, C_{CH3OHd} equal zero.
- (2)(i) NO_{Xmass}=Oxides of nitrogen emissions, in grams per test phase.
- (ii) Density_{NO2}=Density of oxides of nitrogen is 54.16 g/ft³ (1.913 kg/m³) assuming they are in the form of nitrogen dioxide, at 68 °F (20 °C) and 760 mm Hg (101.3kPa) pressure.
- (iii)(A) NOx_{conc} =Oxides of nitrogen concentration of the dilute exhaust sample corrected for background, in ppm.
- (B) $NOx_{conc}=NOx_{e}-NOx_{d}(1-(1/DF))$. Where:

- (iv) NOx_e=Oxides of nitrogen concentration of the dilute exhaust sample as measured, in ppm.
- (v) NOx_d =Oxides of nitrogen concentration of the dilution air as measured, in ppm.
- (3)(i) $\tilde{\text{CO}}_{\text{mass}}$ =Carbon monoxide emissions, in grams per test phase.
- (ii) Density $_{\rm co}=$ Density of carbon monoxide is 32.97 g/ft³ (1.164 kg/m³), at 68 $^{\circ}{\rm F}$ (20 $^{\circ}{\rm C})$ and 760 mm Hg (101.3 kPa) pressure.
- (iii)(A) CO_{conc} =Carbon monoxide concentration of the dilute exhaust sample corrected for background, water vapor, and CO_2 extraction, in ppm.
 - (B) $CO_{conc} = CO_e CO_d(1 (1/DF))$.

Where:

- (iv)(A) CO_e=Carbon monoxide concentration of the dilute exhaust volume corrected for water vapor and carbon dioxide extraction, in ppm.
- (B)
- CO_{e} =(1-0.01925CO_{2e}-0.000323R)CO_{em} for petroleum fuel with hydrogen to carbon ratio of 1.85:1.
- (C) CO_e =[1-(0.01 + 0.005HCR) CO_{2e} -0.000323R] CO_{em} for methanol fuel, where HCR is hydrogento-carbon ratio as measured for the fuel used.
- (v) CO_{em} =Carbon monoxide concentration of the dilute exhaust sample as measured, in ppm.
- (vi) CO_{2e} =Carbon dioxide concentration of the dilute exhaust sample, in percent.
- (vii) R=Relative humidity of the dilution air, in percent (see §86.142(n)).
- (viii)(A) CO_d =Carbon monoxide concentration of the dilution air corrected for water vapor extraction, in ppm.
- (B) $CO_d = (1 0.000323R)CO_{dm}$.

Where:

- (ix) CO_{dm} =Carbon monoxide concentration of the dilution air sample as measured, in ppm.
- NOTE: If a CO instrument which meets the criteria specified in §86.111 is used and the conditioning column has been deleted, CO_{em} must be substituted directly for CO_{e} and CO_{dm} must be substituted directly for CO_{d} .
- (4)(i) CO_{2mass} =Carbon dioxide emissions, in grams per test phase.
- (ii) Density CO₂=Density of carbon dioxide is 51.81 g/ft 3 (1.830 kg/m 3), at 68 $^\circ$ F (20 $^\circ$ C) and 760 mm Hg (101.3 kPa) pressure.

(iii)(A) CO_{2conc}=Carbon dioxide concentration of the dilute exhaust sample corrected for background, in percent.

(B) $CO_{2conc} = CO_{2e} - CO_{2d}(1 - (1/DF))$.

Where:

(iv) CO_{2d} =Carbon dioxide concentration of the dilution air as measured, in percent.

(5)(i) CH₃OH_{mass}=Methanol emissions corrected for background, in grams per test phase.

(ii) Density_{CH3OH}=Density of methanol is 37.71 g/ft³ (1.332 kg/m³), at 68 °F

(20 °C) and 760 mmHg (101.3kPa) pressure

(iii)(A) CH₃OH_{conc}=Methanol concentration of the dilute exhaust corrected for background, ppm.

(B) $CH_3OH_{conc} = C = CH_3OH_e - C_{CH_3OH_d}(1 - (1/DF))$

Where:

(iv)(A) C_{CH3OHe} =Methanol concentration in the dilute exhaust, ppm.

(B)

$$C_{\text{CH3OHe}} = \frac{3.813 \times 10^{-2} \times C_{\text{CH3OHR}} \times T_{\text{EM}} \left[\left(A_{\text{S1}} \times AV_{\text{S1}} \right) + \left(A_{\text{S2}} \times AV_{\text{S2}} \right) \right]}{A_{\text{CH3OHR}} \times P_{\text{B}} \times V_{\text{EM}}}$$

(v)(A) C_{CH3OHd} =Methanol concentration in the dilution air, ppm.

$$C_{CH3OHd} = \frac{3.813\times10^{-2}\times C_{CH3OHR}\times T_{DM}\big[\big(A_{D1}\times AV_{D1}\big) + \big(A_{D2}\times AV_{D2}\big)\big]}{A_{CH3OHR}\times P_{B}\times V_{DM}}$$

(vi) $C_{\text{CH3OHR}} = \text{Concentration of methanol in standard sample for calibration of GC, } \mu g/ml.$

(vii) A_{CH3OHR} =GC peak area of standard sample.

(viii) T_{EM} =Temperature of methanol sample withdrawn from dilute exhaust, ${}^{\circ}\text{R}.$

(ix) $T_{\rm DM}\!\!=\!\! Temperature$ of methanol sample withdrawn from dilution air, $^{\circ}R$

(x) P_B =Barometric pressure during test, mm Hg.

(xi) V_{EM} = V_{Olume} of methanol sample withdrawn from dilute exhaust, ft³.

(xii) V_{DM} =Volume of methanol sample withdrawn from dilution air, ft^3 .

(xiii) A_s =GC peak area of sample drawn from dilute exhaust.

(xiv) A_D =GC peak area of sample drawn from dilution air.

(xv) AV_s=Volume of absorbing reagent (deionized water) in impinger

through which methanol sample from dilute exhaust is drawn, ml.

(xvi) AV_D = Volume of absorbing reagent (deionized water) in impinger through which methanol sample from dilution air is drawn, ml.

(6)(i) HCHO_{mass} = Formaldehyde emissions corrected for background, in grams per test phase.

(ii) Density_{HCHO} = Density of formaldehyde is 35.36 g/ft^3 (1.249 kg/m³), at 68 °F (20 °C) and 760 mmHg (101.3 kPa) pressure.

(iii)(A) HCHO_{conc} = Formaldehyde concentration of the dilute exhaust corrected for background, in ppm.

(B) $HCHO_{conc} = C_{HCHOe} - C_{HCHOd}$ (1 – (1/DF))

Where:

 $\label{eq:contraction} (iv)(A) \ C_{HCHOe} \ = \ Formaldehyde \ concentration in dilute exhaust, in ppm.$

(B

$$C_{HCHOe} = \frac{4.069 \times 10^{-2} \times C_{FDE} \times V_{AE} \times Q \times T_{EF}}{V_{SE} \times P_{B}}$$

(v)(A) C_{HCHOd} = Formaldehyde concentration in dilution air in ppm.

$$C_{HCHOd} = \frac{4.069 \times 10^{-2} \times C_{FDA} \times V_{AA} \times Q \times T_{DF}}{V_{SA} \times P_{B}}$$

(vi) C_{FDE} = Concentration of DNPH derivative of formaldehyde from dilute exhaust sample in sampling solution,

(vii) V_{AE} = Volume of sampling solution for dilute exhaust formaldehyde sample, ml.

(viii)(A) Q = Ratio of molecular weights of formaldehyde to its DNPH derivative.

(B) Q = 0.1429.

(ix) T_{EF} = Temperature of formaldehyde sample withdrawn from dilute ex-

(x) V_{SE} = Volume of formaldehyde sample withdrawn from dilute exhaust,

(xi) P_B = Barometric pressure during test, mm Hg.

(xii) $C_{FDA} = Concentration of DNPH$ derivative of formaldehyde from dilution air sample in sampling solution, μg/ml.

(xiii) V_{AA} = Volume of sampling solution for dilution air formaldehyde sample, ml.

(xiv) T_{DF} = Temperature of formaldehyde sample withdrawn from dilution air, °R.

(xv) V_{SA} = Volume of formaldehyde sample withdrawn from dilution air,

(7)(i) DF = $13.4/[CO_{2e} + (HC_e + CO_e)]$ 10-4] for petroleum-fueled vehicles.

$$DF = \frac{100 \left(\frac{X}{x + y/2 + 3.76(x + y/4 - z/2)} \right)}{CO_{2e} + \left(HC_e + CO_e + C_{CH_3OHe} + C_{HCHOe} \right) 10^{-4}}$$

for methanol-fueled vehicles where fuel composition is $C_x H_y O_z$ as measured for the fuel used.

 $(iii)(A) K_H = Humidity correction fac-$

(B) $K_H = 1/[1 - 0.0047(H - 75)]$. (C) For SI units, $K_H = 1/[1 - 0.0329(H - 1)]$ 10.71)].

Where:

(iv)(A) H = Absolute humidity in grains (grams) of water per pound (kilogram) of dry air.

(B) $H = [(43.478)R_a \times P_d]/[P_B - (P_d \times R_a)]$

(C) For SI units, $H = [(6.211)R_a \times P_d]/$ $[P_B - (P_d \times R_a/100)].$

(v) R_a = Relative humidity of the ambient air, percent.

(vi) P_d = Saturated vapor pressure, mm Hg (kPa) at the ambient dry bulb temperature.

(vii) P_B = Barometric pressure, mm Hg (kPa),

(viii)(A) V_{mix} = Total dilute exhaust volume in cubic feet per test phase corrected to standard conditions (528 °R (293 °K) and 760 mm Hg (101.3 kPa)).

(B) For PDP-CVS, V_{mix} is:

$$V_{mix} = \frac{V_o \times N \times \left(P_B - P_4\right) \times 528}{760 \times T_P}$$

(C) For SI units,

$$V_{\text{mix}} = \frac{V_{\text{o}} \times N \times (P_{\text{B}} - P_{4}) \times 293}{101.3 \times T_{\text{p}}}$$

Where:

(ix) V_o = Volume of gas pumped by the positive displacement pump, in cubic feet (m³) per revolution. This volume is dependent on the pressure differential across the positive displacement pump.

(x) \dot{N} = Number of revolutions of the positive displacement pump during the test phase while samples are being collected.

(xi) P_B = Barometric pressure, mm Hg (kPa).

(xii) P_4 = Pressure depression below atmospheric measured at the inlet to the positive displacement pump, in mm Hg (kPa) (during an idle mode).

(xiii) T_p = Average temperature of dilute exhaust entering positive displacement pump during test, ${}^{\circ}R({}^{\circ}K)$.

(d) For petroleum-fueled vehicles, example calculation of mass values of exhaust emissions using positive displacement pump:

(1) For the "transient" phase of the cold start test assume the following:

 $\begin{array}{c} V_o = 0.29344 \ ft^3/rev; \ N = 10,485; \ R = 48.0 \\ pct; \ R_a = 48.2 \ percent; \ P_B = 762 \ mm \ Hg; \\ P_d = 22.225 \ mm \ Hg; \ P_4 = 70 \ mm \ Hg; \ T_p \\ = 570 \ ^\circ R; \ HC_e = 105.8 \ ppm, \ carbon \ equivalent; \ NO_{xe} = 11.2 \ ppm; \ CO_{em} = 306.6 \\ ppm; \ CO_{2e} = 1.43 \ percent; \ HC_d = 12.1 \\ ppm; \ NO_{xd} = 0.8 \ ppm; \ CO_{dm} = 15.3 \ ppm; \\ CO_{2d} = 0.032 \ percent; \ D_{ct} = 3.598 \ miles. \\ Then: \end{array}$

(i) $V_{\rm mix} = (0.29344)(10,485)(762 - 70)(528)/(760)(570) = 2595.0 \ {\rm ft}^3 \ {\rm per \ test \ phase}.$

(ii) H=(43.478)(48.2)(22.225)/762 – (22.225)(48.2/100) = 62 grains of water per pound of dry air.

(iii) $K_H = 1/[1 - 0.0047(62 - 75)] = 0.9424$

(iv) $CO_e = [1 - 0.01925(1.43) - 0.000323(48)](306.6) = 293.4 \text{ ppm}.$

(v) $CO_d = [1 - 0.000323(48)](15.3) = 15.1$

(vi) DF = 13.4/[1.43 = 10 minus;4(105.8 + 293.4)] = 9.116.

(vii) $HC_{conc} = 105.8 - 12.1(1 - 1/9.116) = 95.03 \text{ ppm}.$

(viii) HC_{mass} = (2595)(16.33)(95.03/1,000,000) = 4.027 grams per test phase.

(ix) $NOx_{conc} = 11.2 - 0.8(1 - 1/9.116) = 10.49 \text{ ppm}.$

(x) NOx_{mass} = (2595)(54.16)(10.49/1000,000)(0.9424) = 1.389 grams per test phase.

(xi) $CO_{conc} = 293.4 - 15.1(1 - 1/9.116) = 280.0$ ppm.

(xii) $CO_{mass} = (2595)(32.97)(280/1,000,000)$ = 23.96 grams per test phase.

(xiii) $CO_{2conc} = 1.43 - 0.032(1 - 1/9.116)$ = 1.402 percent.

(xiv) $CO_{2mass} = (2595.0)(51.85)(1.402/100)$ = 1886 grams per test phase.

(2) For the stabilized portion of the cold start test assume that similar calculations resulted in the following:

(i) $HC_{mass} = 0.62$ grams per test phase. (ii) $NOx_{mass} = 1.27$ grams per test

phase. (iii) $CO_{mass} = 5.98$ grams per test phase.

(iv) $CO_{2mass} = 2346$ grams per test phase.

(v) $D_s = 3.902$ miles.

(3) For the "transient" portion of the hot start test assume that similar calculations resulted in the following:

(i) $HC_{mass} = 0.51$ gram per test phase.

(ii) $NOx_{mass} = 1.38$ grams per test phase.

(iii) $CO_{mass} = 5.01$ grams per test phase.

(iv) $CO_{2mass} = 1758$ grams per test phase.

(v) $D_{ht} = 3.598$ miles.

(4) Weighted mass emission results:

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(i) HC_{wm} = 0.43[(4.027 + 0.62)/(3.598 + 3.902)] + 0.57[(0.51 + 0.62)/(3.598 + 3.902)] = 0.352 grams per vehicle mile.
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(ii) $NO_{xwm} = 0.43[(1.389 + 1.27)/(3.598 + 3.902)] + 0.57[(1.38 + 1.27)/(3.598 + 3.902)] = 0.354$ grams per vehicle mile.

(iii) $CO_{wm} = 0.43[(23.96 + 5.98)/(3.598 + 3.902)] = 0.57[(5.01 + 5.98)/(3.598 + 3.902)] = 2.55$ grams per vehicle mile.

(iv) $CO_{2wm} = 0.43[(1886 + 2346)/(3.598 + 3.902) = 0.57[(1758 + 2346)/(3.598 + 3.902)] = 555$ grams per vehicle mile.

(e) For methanol-fueled vehicle with measured fuel composition of $CH_{3.14}$ $O_{0.6}$ example calculation of exhaust emissions using positive displacement pump:

(1) For the "transient" phase of the cold start test assume the following:

 $V_0 = 0.29344 \text{ ft}^3/\text{rev}; N = 10,485; R = 48.0$ pct; $R_a = 48.2$ percent; $P_B = 762$ mm Hg; $P_d = 22.225 \text{ mm Hg}; P_4 = 70 \text{ mm Hg}; T_p$ 570 °R; FID HC_e = 81.6 ppm, carbon equivalent; r = 0.75; $C_{CH30HR} = 71 \text{ ug/ml}$; $T_{EM} = 567 \, {}^{\circ}R; \, A_{CH3OHR} = 3660; \, V_{EM} \, 1.18$ ft^3 ; $A_{S1} = 4460$; $AV_{s1} = 25.2$ ml; $A_{s2} = 360$; $AV_{s2} = 24.9 \text{ ml}$; $T_{DM} = 532 \, {}^{\circ}\text{R}$; $V_{DM} = 1.17$ $ft^3;\ A_{D1}=\ 110;\ AV_{D1}=\ 25.0\ ml;\ A_{D2}=\ 10;$ $AV_{D2} = 25.1 \text{ ml}; C_{FDE} = 20 \text{ ug/ml}; V_{AE} =$ 5.0 ml; Q = 0.1429; $T_{EF} = 569 \, ^{\circ}\text{K}$; $V_{SE} = 0.30$ ft^3 ; $C_{FDA} = 1 \text{ ug/ml}$; $V_{AA} = 5.0 \text{ ml}$; $T_{DF} =$ 532 °R; $V_{SA} = 0.31$ ft³; $NOx_e = 11.2$ ppm; $CO_{em} = 306.6 \text{ ppm}$; $CO_{2e} = 1.43 \text{ pct}$; FID $HC_d = 12.1 \text{ ppm}$; $NOx_d = 0.8 \text{ ppm}$; $CO_{dm} =$ 15.3 ppm; $\hat{CO}_{2d} = 0.032$ percent; $\hat{D}_{ct} =$ 3.598 miles.

Then:

(i) $V_{mix} = (0.29344)(10,485)(762 - 70)(528)/(760)(570) = 2595.0 \text{ ft}^3 \text{ per test phase.}$

(ii) H = $(43.478)(48.2)(22.225)/(762 - (22.225 \times 48.2/100))$ = 62 grains of water per pound of dry air.

(iii) $K_H = 1/[1 - 0.0047(62 - 75)] = 0.9424$.

(iv) $CO_e = [1 - (0.01 + 0.005 \times 3.14 \times 1.43) - 0.000323(48))] \times 306.6 = 291.9 \text{ ppm.}$

(v) $CO_d = (1 - 0.000323(48)) \times 15.3 = 15.1$ ppm.

 $\begin{array}{lll} \text{(vi)} & C_{\text{CH3OHe}} & = \\ (3.813\times10^{-2})(71)(567)[(4460)(25.2) & + \\ (360)(24.9)] \ / \ (3660)(762)(1.18) = 56.60 \ \text{ppm.} \\ \text{(vii)} \ \ \text{DF} & = 100(1/[1 + (3.14/2) + 3.76(1 + (3.14/4) - (0.6/2))])/1.43 + 10^{-4}[(81.6 + 291.9 + (1 - 0.75)(56.60)] + 8.350 \end{array}$

(viii) C_{CH3OHd} = (3.813×10⁻²)(71)(532)[(110)(25.0) (10)(25.1)] / (3660)(762)(1.17) = 1.32 ppm.

(ix) $CH_3OH_{conc} = 56.60 - 1.32(1 - 1/8.350) = 55.44$ ppm.

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(x) CH_3OH_{mass} = 2595.0 \times 37.71 \times (55.44/1,000,000) = 5.43 grams per test phase.
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(xi) HC_{conc} [81.6 - (0.75) (56.60)] - [12.1 - (0.75) (1.32)] (1-1/8.530) = 29.34 ppm.

(xii) $HC_{mass}=(2594)$ (16.33) (29.34/1,000,000)=1.24 grams per test phase.

(xiii) $C_{HCHOe} = 4.069 \times 10^{-2}(20)(5)(0.1429)(569)/(0.30)(762) = 1.4473$ ppm.

(xiv) $C_{HCHOd} = 4.069 \times 10^{-2}(1)(5)(0.1429)(532)/(0.31)(762) = 0.0655$ ppm.

(xv) HCHO_{conc}=1.4473 – 0.0655(1 – 1/8.350)=1.3896 ppm.

(xvi) HCHO_{mass}=(2595)(35.36)(1.3896/1,000,000)=0.1275 grams per test phase.

(xvii) THCE = 1.24+(13.8756/32.042)(5.43) + (13.8756/30.0262)(0.1275) = 3.65 grams per test phase.

(xviii) NO x_{conc} =11.2 - (0.8) (1 - 1/8.350)=10.50 ppm

 $\begin{array}{cccc} (xix) & NOx_{mass} = (2595)(54.16)(10.50/1,000,000)(0.9424) = 1.390 & grams & per & test \\ phase. & & & & & & & & & \\ \end{array}$

(xx) CO_{conc}=291.9-15.1 (1-1/8.350)=278.61 ppm.

(xxi) $\dot{\text{CO}}_{\text{mass}}$ =(2595.0)(32.97)(278.69/1,000,000)=23.84 grams per test phase.

(xxii) $CO_{2conc}=1.43-0.032$ (1-1/8.350)=1.402 percent.

(xxiii) CO_{2mass} =(2595.0)(51.85)(1.402/100)=1886 grams per test phase.

(2) For the stabilized portion of the cold start test assume that similar calculations resulted in the following:

(i) THCE=0.55 grams per test phase.

(ii) NOx_{mass} =1.27 grams per test phase. (iii) CO_{mass} =5.98 grams per test phase.

(iv) CO_{2mass} =2346 grams per test phase.

(v) $D_s=3.902$ miles.

(3) For the "transient" portion of the hot start test assume that similar calculations resulted in the following:

(i) THCE=0.67 grams as carbon equivalent per test phase.

(ii) NOx_{mass} =1.38 grams per test phase.

(iii) CO_{mass} =5.01 grams per test phase. (iv) CO_{2mass} =1758 grams per test phase.

(iv) $CO_{2\text{mass}} = 1736$ grams per test (v) $D_{\text{ht}} = 3.598$ miles.

(4) Weighted emission results:

(i) THCE_{wm}=0.43[(3.65+0.55)/(3.598+3.902)]+0.57[(0.67+0.55)/(0

(3.598+3.902)]=0.334 grams as carbon equivalent per mile.

(ii) NOx_{wm} =0.43[(1.390+1.27)/(3.598+3.902)]+0.57[1.38+1.27)/(3.598+3.902)]=0.354 grams per vehicle mile.

(iii) CO_{wm}=0.43[(23.84+5.98)/ (3.598+3.902)]+0.57[(5.01+5.98)/ (3.598+3.902)]-2.54 grams per vehicle

(3.598+3.902)]=2.54 grams per vehicle mile.

(iv) CO_{2wm} +0.43[(1886+2346)/(3.598+3.902)]+0.57[(1758+2346)/

(3.598+3.902)]=555 grams per vehicle mile.

[54 FR 14535, Apr. 11, 1989, as amended at 59 FR 39649, Aug. 3, 1994]

§86.144-94 Calculations; exhaust emissions.

The final reported test results shall be computed by use of the following formula:

(a) For light-duty vehicles and light duty trucks:

$$Y_{wm} = 0.43 \left(\frac{(Y_{ct} + Y_s)}{(D_{ct} + D_s)} \right) + 0.57 \left(\frac{(Y_{ht} + Y_s)}{(D_{ht} + D_s)} \right)$$

Where:

- (1) Y_{WM} = Weighted mass emissions of each pollutant, *i.e.*, THC, CO, THCE, NMHC, NMHCE, CH₄, NO_X, or CO₂, in grams per vehicle mile.
- (2) \dot{Y}_{ct} =Mass emissions as calculated from the "transient" phase of the cold start test, in grams per test phase.
- (3) Y_{ht}=Mass emissions as calculated from the "transient" phase of the hot start test, in grams per test phase.
- (4) Y_s =Mass emissions as calculated from the ''stabilized'' phase of the cold start test, in grams per test phase.
- (5) D_{ct} =The measured driving distance from the ''transient'' phase of the cold start test, in miles.
- (6) D_{ht} =The measured distance from the ''transient'' phase of the hot start test, in miles.
- (7) D_s =The measured driving distance from the ''stabilized'' phase of the cold start test, in miles.
- (b) The mass of each pollutant for each phase of both the cold start test and the hot start test is determined from the following:
 - (1) Total hydrocarbon mass:

 $HC_{mass}{=}V_{mix} \times Density_{HC} \times (HC_{cone}/1,000,000)$

(2) Oxides of nitrogen mass:

 $\begin{array}{cccc} NOx_{mass} = & V_{mix} \times Density_{NO2} \times K_{H} \times \\ & (NOx_{conc}/1,000,000) \end{array}$

(3) Carbon monoxide mass:

 $CO_{mass} = V_{mix} \times Density_{CO} \times (CO_{conc} / 1.000.000)$

(4) Carbon dioxide mass:

 $CO_{2mass} = V_{mix} \times Density_{CO2} \times (CO_{2conc}/100)$

(5) Methanol mass:

- $CH_3OH_{mass}=V_{mix} \times Density_{CH}3OH > (CH_3OH_{conc}/1,000,000)$
 - (6) Formaldehyde mass:
- $\begin{array}{ccc} \text{HCHO}_{\text{mass}} = \text{V}_{\text{mix}} & \times & \text{Density}_{\text{HCHO}} \\ & (\text{HCHO}_{\text{conc}} / 1,000,000) \end{array}$
- (7) Total hydrocarbon equivalent mass:

 $\begin{array}{ccccccccc} THCE_{mass} &=& HC_{mass} &+& 13.8756/32.042 &\times \\ &(CH_3OH_{mass}) &+& 13.8756/32.0262 &\times \\ &(HCHO_{mass}) & & \end{array}$

- (8) Non-methane hydrocarbon mass:
- $\begin{array}{lll} NMHC_{mass} &= V_{mix} \times Density_{NMHC} \times \\ & (NMHC_{conc}/1,000,000) \end{array}$
- (9) Non-methane hydrocarbon equivalent mass:

 $\begin{array}{lll} NMHCE_{mass} = NMHC_{mass} + 13.8756/32.042 \times \\ (CH_{3}OH_{mass}) & + & 13.8756/30.0262 & \times \\ (HCHO_{mass}) & & \end{array}$

(10) Methane mass:

 $CH_{4mass}=V_{mix}=Density_{CH4}=(CH_{4conc}/1,000,00)$

- (c) Meaning of symbols:
- (1)(i) HC_{mass}=Total hydrocarbon emissions, in grams per test phase.
- (ii) $Density_{HC}$ =Density of total hydrocarbon.
- (A) For gasoline-fuel, diesel-fuel and methanol fuel; Density $_{\rm HC}$ =16.33 g/ft³-carbon atom (0.5768 kg/m³-carbon atom), assuming an average carbon to hydrogen ratio of 1:1.85, at 68 °F (20 °C) and 760 mm Hg (101.3 kPa) pressure.
- (B) For natural gas and liquefied petroleum gas-fuel; Density_HC=1.1771 (12.011+H/C (1.008)) g/ft³-carbon atom (0.04157(12.011+H/C (1.008))kg/m³-carbon atom), where H/C is the hydrogen to carbon ratio of the hydrocarbon components of the test fuel, at 68 °F (20 °C) and 760 mm Hg (101.3 kPa) pressure.